

# Torquoselective Ring Closures of Chiral Amido Trienes Derived from Allenamides. A Tandem Allene Isomerization—Pericyclic Ring-Closure—Intramolecular Diels—Alder Cycloaddition

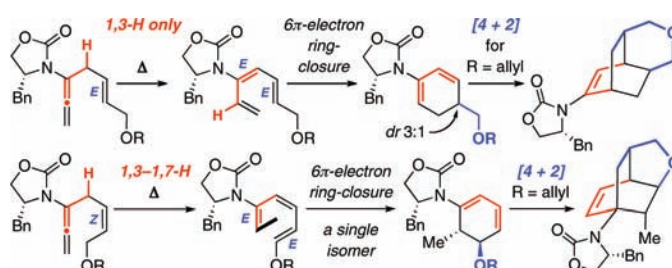
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Received December 7, 2009

## ABSTRACT



A new torquoselective ring-closure of chiral amide-substituted 1,3,5-hexatrienes and its application in tandem with [4 + 2] cycloaddition are described. The trienes were derived via either a 1,3-H or 1,3-H–1,7-H shift of  $\alpha$ -substituted allenamides, and the entire sequence through the [4 + 2] cycloaddition could be in tandem from allenamides.

We recently reported isomerizations of  $\alpha$ -substituted allenamides **1** to give amido dienes **2** via stereoselective 1,3-H shift (Scheme 1).<sup>1</sup> In addition, with R = vinyl, the resulting 1,3,5-hexatrienes **2'** were found to be well suited for a  $6\pi$ -electron electrocyclic ring closure that could be in tandem with the 1,3-H shift, leading to novel chiral cyclic amido dienes **3**<sup>2,3</sup> directly from allenamides.<sup>4,5</sup> The rapid access of 1,3,5-hexatrienes via a simple isomerization of allenes<sup>6–8</sup> allowed us to envision a new torquoselective ring-closure<sup>9–13</sup>

involving chiral amido trienes **4**. This asymmetric transformation could potentially lead to a remote 1,6-stereochemical induction while affording cyclic amido dienes **5**, which should be useful for cycloadditions. We communicate here this torquoselective process and its application in tandem with [4 + 2] cycloadditions.

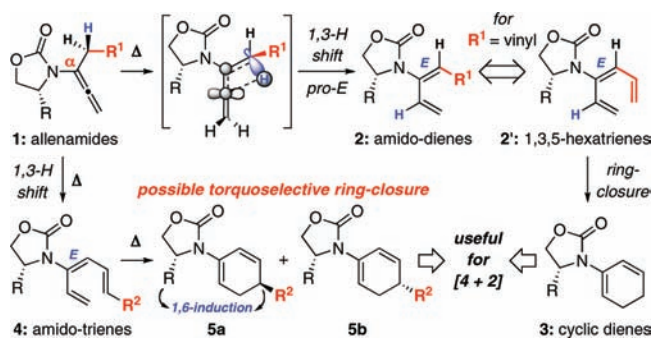
(1) Hayashi, R.; Hsung, R. P.; Feltenberger, J. B.; Lohse, A. G. *Org. Lett.* **2009**, *11*, 2125.

(2) For reviews on chemistry of dienamides, see: (a) Overman, L. E. *Acc. Chem. Res.* **1980**, *13*, 218. (b) Petrzilka, M. *Synthesis* **1981**, 753. (c) Campbell, A. L.; Lenz, G. R. *Synthesis* **1987**, 421. Also see: (d) Krohn, K. *Angew. Chem., Int. Ed.* **1993**, *32*, 1582. (e) Enders, D.; Meyer, O. *Liebigs Ann.* **1996**, 1023.

(3) (a) For a review on the synthesis of enamides, see: Tracey, M. R.; Hsung, R. P.; Antoline, J.; Kurtz, K. C. M.; Shen, L.; Slafer, B. W.; Zhang, Y. In *Science of Synthesis, Houben-Weyl Methods of Molecular Transformations*; Weinreb, S. M., Ed.; Georg Thieme Verlag: Stuttgart, 2005; Chapter 21.4. (b) For a leading review on recent chemistry of enamides, see: Carbery, D. R. *Org. Biomol. Chem.* **2008**, *9*, 3455. (c) Rappoport, Z. *The Chemistry of Enamines in The Chemistry of Functional Groups*; John Wiley and Sons: New York, 1994.

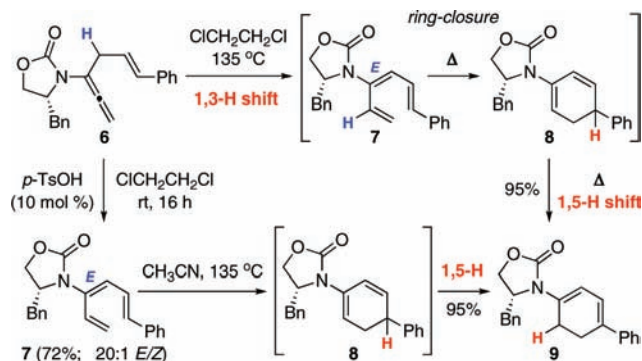
(4) For a leading review on allenamide chemistry, see: Hsung, R. P.; Wei, L.-L.; Xiong, H. *Acc. Chem. Res.* **2003**, *36*, 773.

### Scheme 1. New Torquoselective Pericyclic Ring Closure



Our intention was quickly met with two unexpected findings. Initially, when heating allenamide **6**<sup>14,15</sup> in  $\text{ClCH}_2\text{CH}_2\text{Cl}$  at  $135^\circ\text{C}$ , instead of isolating the desired amido diene **8** from ring closure of the triene **7**, we found **9**<sup>16</sup> in almost quantitative yield, thereby implying a 1,5-H shift had taken place (Scheme 2). Similar results were attained when using triene **7** generated from **6** via an acid-promoted 1,3-H shift using 10 mol % of *p*-TsOH.

### Scheme 2. Complication with the 1,5-H Shift into Conjugation



We quickly made a minor substrate adjustment to prevent such 1,5-H shift, but that led to the second unexpected finding as shown in Scheme 3. Heating  $\alpha$ -prenylated allenamide **10a** led to a mixture of two ring-closure products: the desired 2-amido diene **13a** and the unexpected 1-amido diene **14a**

(5) For recent reports on allenamide chemistry in 2009, see: (a) Hashimoto, K.; Horino, Y.; Kuroda, S. *Heterocycles* **2010**, *80*, 187. (b) Persson, A. K. Å.; Johnston, E. V.; Bäckvall, J.-E. *Org. Lett.* **2009**, *11*, 3817. (c) Skucas, E.; Zbieg, J. R.; Krische, M. J. *J. Am. Chem. Soc.* **2009**, *131*, 5054. (d) Armstrong, A.; Emmerson, D. P. G. *Org. Lett.* **2009**, *11*, 1547. (e) Beccalli, E. M.; Broggin, G.; Clerici, F.; Galli, S.; Kammerer, C.; Rigamonti, M.; Sottocornola, S. *Org. Lett.* **2009**, *11*, 1563. (f) Broggin, G.; Galli, S.; Rigamonti, M.; Sottocornola, S.; Zecchi, G. *Tetrahedron Lett.* **2009**, *50*, 1447. (g) Lohse, A. G.; Hsung, R. P. *Org. Lett.* **2009**, *11*, 3430. (h) Lu, T.; Hayashi, R.; Hsung, R. P.; DeKorver, K. A.; Lohse, A. G.; Song, Z.; Tang, Y. *Org. Biomol. Chem.* **2009**, *9*, 3331. (i) Kimber, M. C. *Org. Lett.* **2010**, DOI: 10.1021/ol1001494.

(6) For general reviews on allenes, see: Krause, N.; Hashmi, A. S. K. *Modern Allene Chemistry*; Wiley-VCH Verlag: Weinheim, 2004; Vols. 1 and 2.

### Scheme 3. Unexpected Competing 1,7-H Shift

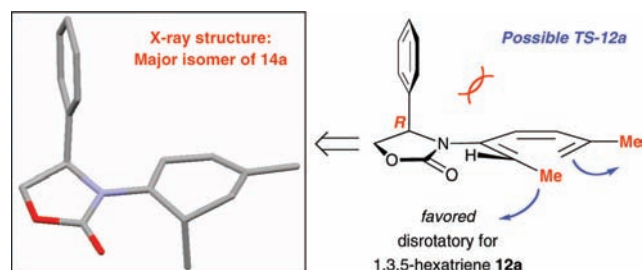
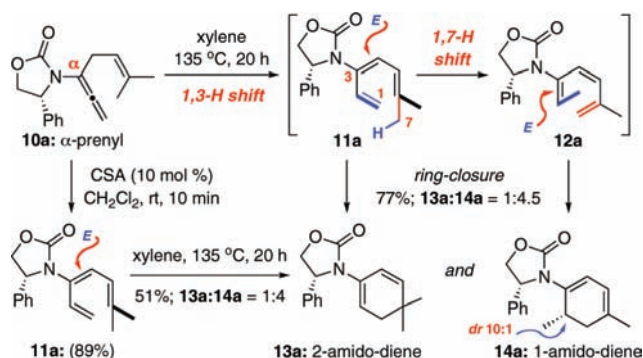


Figure 1. Torquoselective disrotatory ring closure.

in 1:4.5 ratio with **14a** being a 10:1 diastereomeric mixture. The latter implied the presence of amido triene **12a**, which could be rationalized through an antarafacial 1,7-H shift<sup>9,17</sup> from the initial amido triene **11a**, via the methyl group (in red) *syn* to the terminal olefin (in blue). More importantly, stereochemistry for the major isomer of **14a** could be assigned using its single-crystal X-ray structure (Figure 1). This unambiguous assignment suggests that a favored disrotatory course could proceed through the transition state as shown for amido triene **12a**.

(7) For some examples of thermal allene isomerizations, see: (a) Crandall, J. K.; Paulson, D. R. *J. Am. Chem. Soc.* **1966**, *88*, 4302. (b) Bloch, R.; Percec, P. L.; Conia, J.-M. *Angew. Chem., Int. Ed.* **1970**, *9*, 798. (c) Jones, M.; Hendrick, M. E.; Hardie, J. A. *J. Org. Chem.* **1971**, *36*, 3061. (d) Patrick, T. B.; Haynie, E. C.; Probat, W. J. *Tetrahedron Lett.* **1971**, *27*, 423. (e) Lehigh, F.; Hopf, H. *Tetrahedron Lett.* **1987**, *28*, 2697. (f) Meier, H.; Schmitt, M. *Tetrahedron Lett.* **1989**, 5873.

(8) For examples of allenamide isomerizations, see: (a) Overman, L. E.; Clizbe, L. A.; Freerks, R. L.; Marlowe, C. K. *J. Am. Chem. Soc.* **1981**, *103*, 2807. Also see: (b) Farmer, M. L.; Billups, W. E.; Greenlee, R. B.; Kurtz, A. N. *J. Org. Chem.* **1966**, *31*, 2885. For an allenamide isomerization via Grubb's catalyst, see: (c) Kinderman, S. S.; van Maarseveen, J. H.; Schoemaker, H. E.; Hiemstra, H.; Rutjes, F. P. J. T. *Org. Lett.* **2001**, *3*, 2045. (d) Trost, B. M.; Stiles, D. T. *Org. Lett.* **2005**, *7*, 2117.

(9) For reviews for pericyclic ring closures, see: (a) Marvell, E. N. *Thermal Electrocyclic Reactions*; Academic Press: New York, 1980. (b) Okamura, W. H.; de Lera, A. R. In *Comprehensive Organic Synthesis*, Trost, B. M., Fleming, I., Paquette, L. A., Eds.; Pergamon Press: New York, 1991; Vol. 5, pp 699–750. For reviews on ring-closure in natural product synthesis, see: (c) Pindur, U.; Schneider, G. H. *Chem. Soc. Rev.* **1994**, 409. (d) Beaudry, C. M.; Malerich, J. P.; Trauner, D. *Chem. Rev.* **2005**, *105*, 4757.

(10) For examples, see: (a) Martínez, R.; Jiménez-Vázquez, H. A.; Delgado, F.; Tamariz, J. *Tetrahedron* **2003**, *59*, 481. (b) Wallace, D. J.; Klauber, D. J.; Chen, C. Y.; Volante, R. P. *Org. Chem.* **2003**, *5*, 4749. (c) Wabnitz, T. C.; Yu, J.-Q.; Spencer, J. B. *Chem.—Eur. J.* **2004**, *10*, 484.

**Table 1.** Tandem 1,3-H–1,7-H Shift Pericyclic Ring Closure

entry	$\alpha$ -prenyl allenamides	amido-trienes <sup>a</sup>	yield [%]: <sup>c</sup> 13	14 [dr] <sup>d</sup>
1	<b>10b</b> : R <sup>1</sup> = Bn R <sup>2</sup> = R <sup>3</sup> = H	<b>11b</b> [79]	-	50 [15:1]
2	<b>10c</b> : R <sup>1</sup> = <i>i</i> -Pr R <sup>2</sup> = R <sup>3</sup> = H	<b>11c</b> [91]	$\Delta$ in xylene <sup>b</sup>	50 [15:1]
3	<b>10d</b> : R <sup>1</sup> = Me R <sup>2</sup> = Ph; R <sup>3</sup> = H	<b>11d</b> [74]	1,7-H shift–ring-closure	54 [9:1]
4	<b>10e</b> : R <sup>1</sup> = Ph R <sup>2</sup> = R <sup>3</sup> = Ph	<b>11e</b> [86]	1,7-H shift–ring-closure	15 45 [6:1]
5	<b>10b</b> : R <sup>1</sup> = Bn R <sup>2</sup> = R <sup>3</sup> = H		$\Delta$ in xylene <sup>b</sup>	52 [15:1]
6	<b>10c</b> : R <sup>1</sup> = <i>i</i> -Pr R <sup>2</sup> = R <sup>3</sup> = H		$\Delta$ in xylene <sup>b</sup>	62 [15:1]
7	<b>10d</b> : R <sup>1</sup> = Me R <sup>2</sup> = Ph; R <sup>3</sup> = H		tandem 1,3-H–1,7-H shift	21 38 [9:1]
8	<b>10e</b> : R <sup>1</sup> = Ph R <sup>2</sup> = R <sup>3</sup> = Ph		–ring-closure	11 34 [6:1]

<sup>a</sup> All 3-amido trienes **11b–e** were attained from the respective allenamides **10b–e** via 1,3-H shift promoted by CSA (10 mol %). Reactions were run in CH<sub>2</sub>Cl<sub>2</sub> at rt for 10 min, and isolated yields are shown in the respective brackets. <sup>b</sup> For all entries: conc = 0.10 M; temp = 135 °C; time = 20 h. <sup>c</sup> Isolated yields. <sup>d</sup> Ratios determined by <sup>1</sup>H and/or <sup>13</sup>C NMR.

The tandem 1,3-H-1,7-H shift appears to be general whether commencing from amido trienes **11b–e** (entries 1–4 in Table 1) or directly from  $\alpha$ -prenylated allenamides **10b–e** (entries 5–8). It is noteworthy that in the case of allenamide **10b** and **10c** cyclic amido dienes **14b** and **14c** were the only products resulting from a tandem 1,3-H–1,7-H shift–pericyclic ring closure. In addition, there was no equilibration between **13** and **14** under the reaction conditions.<sup>18</sup>

(11) For some examples on recent 6- $\pi$ -electron electrocyclic ring closures of 1,3,5-hexatrienes, see: (a) Bishop, L. M.; Barbarow, J. E.; Bergmen, R. G.; Trauner, D. *Angew. Chem., Int. Ed.* **2008**, *47*, 8100. (b) Sofiyev, V.; Navarro, G.; Trauner, D. *Org. Lett.* **2008**, *10*, 149. (c) Kan, S. B. J.; Anderson, E. A. *Org. Lett.* **2008**, *10*, 2323. (d) Hulot, C.; Blong, G.; Suffert, J. *J. Am. Chem. Soc.* **2008**, *130*, 5046. (e) Benson, C. L.; West, F. G. *Org. Lett.* **2007**, *9*, 2545. (f) Pouwer, R. H.; Schill, H.; Williams, C. M.; Bernhardt, P. V. *Eur. J. Org. Chem.* **2007**, 4699. (g) Jung, M. E.; Min, S.-J. *Tetrahedron* **2007**, *63*, 3682.

(12) For recent work on accelerated ring closures of 1,3,5-hexatrienes, see: (a) Barluenga, J.; Merino, I.; Palacios, F. *Tetrahedron Lett.* **1990**, *31*, 6713. (b) Magomedov, N. A.; Ruggiero, P. L.; Tang, Y. *J. Am. Chem. Soc.* **2004**, *126*, 1624. (c) Tessier, P. E.; Nguyen, N.; Clay, M. D.; Fallis, A. G. *J. Am. Chem. Soc.* **2006**, *128*, 4946. (d) Huntley, R. J.; Funk, R. L. *Org. Lett.* **2006**, *8*, 3403. (e) Yu, T.-Q.; Fu, Y.; Liu, L.; Guo, Q.-X. *J. Org. Chem.* **2006**, *71*, 6157. (f) Sünemann, H. W.; Banwell, M. G.; de Meijere, A. *Eur. J. Org. Chem.* **2007**, 3879.

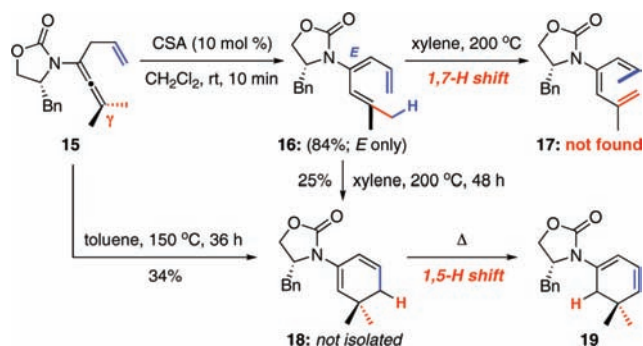
(13) For theoretical studies on substituent effects on electrocyclic ring closure of 1,3,5-hexatrienes, see: (a) Spangler, C. W.; Jondahl, T. P.; Spangler, B. *J. Org. Chem.* **1973**, *38*, 2478. (b) Guner, V. A.; Houk, K. N.; Davies, I. A. *J. Org. Chem.* **2004**, *69*, 8024. (c) Yu, T.-Q.; Fu, Y.; Liu, L.; Guo, G. X. *J. Org. Chem.* **2006**, *71*, 6157. (d) Duncan, J. A.; Calkins, D. E. G.; Chavarha, M. *J. Am. Chem. Soc.* **2008**, *130*, 6740.

(14) (a) Trost, B. M.; Stiles, D. T. *Org. Lett.* **2005**, *7*, 2117. (b) Shen, L.; Hsung, R. P.; Zhang, Y.; Antoline, J. E.; Zhang, X. *Org. Lett.* **2005**, *7*, 3081.

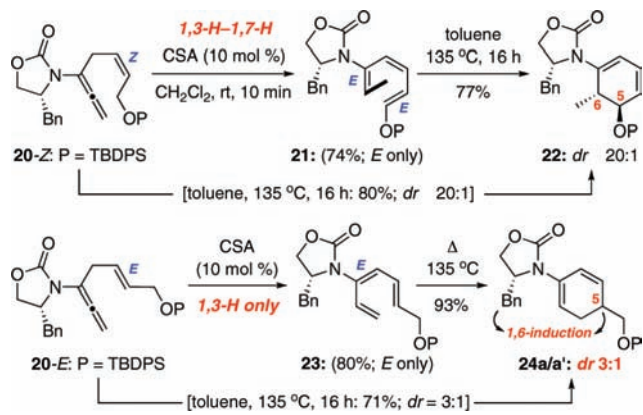
(15) Xiong, H.; Hsung, R. P.; Wei, L.-L.; Berry, C. R.; Mulder, J. A.; Stockwell, B. *Org. Lett.* **2000**, *2*, 2869.

(16) See the Supporting Information.

(17) For recent examples in thermal antarafacial 1,7-H shift of 1,3,5-hexatrienes, see: (a) Kerr, D. J.; Willis, A. C.; Flynn, B. L. *Org. Lett.* **2004**, *6*, 457. (b) Mousavipour, S. H.; Fernández-Ramos, A.; Meana-Pañeda, R.; Martínez-Núñez, E.; Vázquez, S. A.; Ríos, M. A. *J. Phys. Chem. A* **2007**, *111*, 719. (c) Gu, Z.; Ma, S. *Chem.–Eur. J.* **2008**, *14*, 2453. (d) Shu, X.-Z.; Ji, K.-G.; Zhao, S.-C.; Zheng, Z.-J.; Chen, J.; Lu, L.; Liu, X.-Y.; Liang, Y.-M. *Chem.–Eur. J.* **2008**, *14*, 10556.

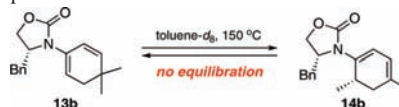
**Scheme 4.** Directional Preference in the 1,7-H Shift

Probing further mechanistically, we found two phenomena associated with this 1,7-H shift process. First, heating either triene **16**, derived from  $\gamma$ -substituted allenamido to no observable 1,7-H shift with cyclic diene **19** as the only identifiable product. Isolation of **19** implies the ring closure of triene **16** had taken place to give **18** followed by 1,5-H shift (Scheme 4). This experiment suggests that there exists a directional preference for the 1,7-H shift in these amido trienes and that it is not sufficient simply having a methyl group (in red) at one terminus of the triene *syn* to the other terminus (in blue).

**Scheme 5.** Diverging Tandem Pathway: *E*- vs *Z*-Olefin

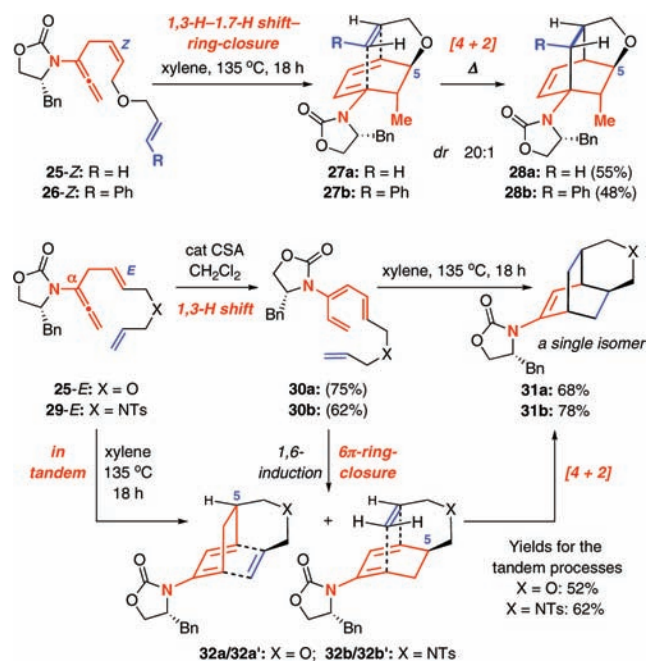
Second, we found a distinct dependence of the 1,7-H shift on the olefinic geometry. As shown in Scheme 5, reactions of allenamides **20-Z** and **20-E** proceeded through distinctly different tandem pathways. While 1,3-H–1,7-H shift occurred with **20-Z** en route to cyclic amido diene **22** via pericyclic ring closure of triene **21**, the reaction of

(18) Independent heating of **13b** or **14b** led to no observable amount of the other compound.



allenamide **20-E** gave **24** with no observable 1,7-H shift. Relative stereochemistry in **22** was assigned on the basis of a disrotatory ring closure, while the absolute stereochemistry was assessed on the basis of **14a**. Cyclic amido diene **24** was found as a 3:1 isomeric mixture with respect to C5, thereby implying that albeit modest and unassigned at this time,<sup>19</sup> a rather impressive 1,6-asymmetric induction took place during the ring closure.

**Scheme 6.** In Tandem with [4 + 2] Cycloaddition

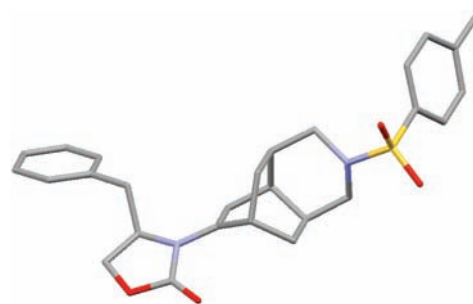


To both accentuate this dichotomy and render these stereoselective ring closures synthetically useful, we embarked on tandem processes that would include [4 + 2] cycloadditions. As shown in Scheme 6, reactions of allenamides **25-Z** and **26-Z** led to tricycles **28a** and **28b** as a single isomer through a highly stereoselective [4 + 2] cycloaddition of cyclic amido dienes **27a** and **27b**, respectively, thereby constituting a quadruple tandem process of 1,3-H–1,7-H shift–6 $\pi$ -electron pericyclic ring-closure–[4 + 2] cycloaddition. Stereochemical outcome of the cycloaddition was controlled through the C5-stereocenter, which was installed during the torquoselective ring closure.

(19) Attempts were made to attain an X-ray but were not successful mainly because of difficulties in the separation of diastereomers. Further efforts are ongoing.

(20) Unfortunately, Diels–Alder cycloadditions of **32a/a'** or **32b/b'** in Scheme 6 did not provide conclusive insight into the stereochemical outcome of ring closures of **30a** and **30b**.

In contrast, reactions of allenamides **25-E** (for a direct comparison with **25-Z**) and **29-E** led the respective tricycles **31a** and **31b** (assigned via X-ray: see Figure 2) in excellent yields and high diastereoselectivity proceeding from amido-trienes **30a** and **30b** or directly from the allenamides in a triple tandem process. In either case, the [4 + 2] cycloaddition presumably went through the diastereomeric pairs **32a/a'** (X = O) and **32b/b'** (X = NTs) given the modest 1,6-induction found for **20-E**. It is noteworthy that despite not having assigned these diastereomers with ratios undermined,<sup>20</sup> both pairs would actually converge to give **31a** and **31b**, respectively. These tandem processes provide a rapid assembly of complex tricycles from very simple allenamides, thereby manifesting their tremendous power and synthetic potential.



**Figure 2.** X-ray structure of **31b**.

We have described here a new torquoselective ring closure of chiral amide-substituted 1,3,5-hexatrienes and its application in tandem with [4 + 2] cycloaddition. The 1,3,5-hexatrienes were derived via either a 1,3-H or 1,3-H–1,7-H shift of  $\alpha$ -substituted allenamides, and the entire sequence through the Diels–Alder could be in tandem from allenamides. Applications of these new tandem processes as well as mechanistic understanding and improvement of the observed 1,6-asymmetric induction are underway.

**Acknowledgment.** We thank the NIH (GM066055) for support and Dr. Victor Young (University of Minnesota) for X-ray structural analysis.

**Supporting Information Available:** Experimental procedures as well as NMR spectra, characterizations, and X-ray structural files. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL902821W